



Knowledge that will change your world

The Chemistry of the metabolome

Stephen Barnes, PhD
University of Alabama at Birmingham
sbarnes@uab.edu

What is a component of the metabolome?

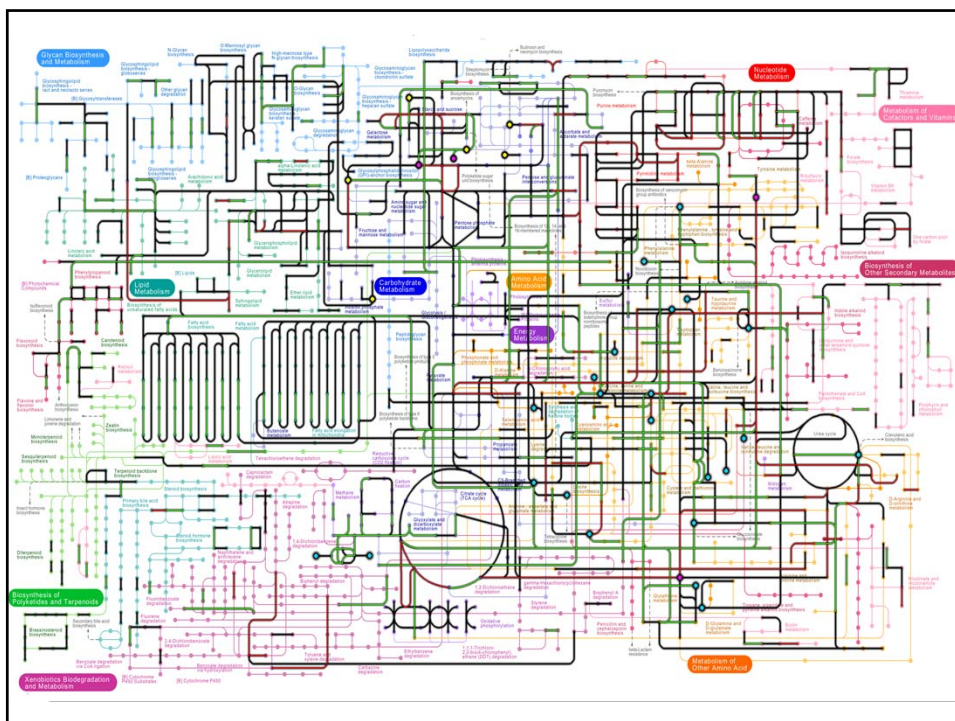
- In the context of metabolomics, it is *compound of any origin that has a molecular weight <1,500 Da that can be detected in the biological system being studied*
- This is an arbitrary definition

The metabolome is more than what's in textbooks

Metabolites synthesized from small molecule precursors by human cells

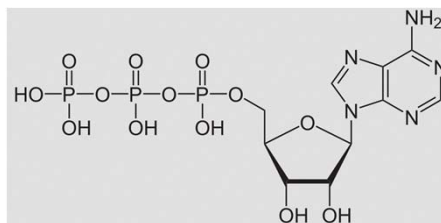


Metabolite pool in tissues and biofluids

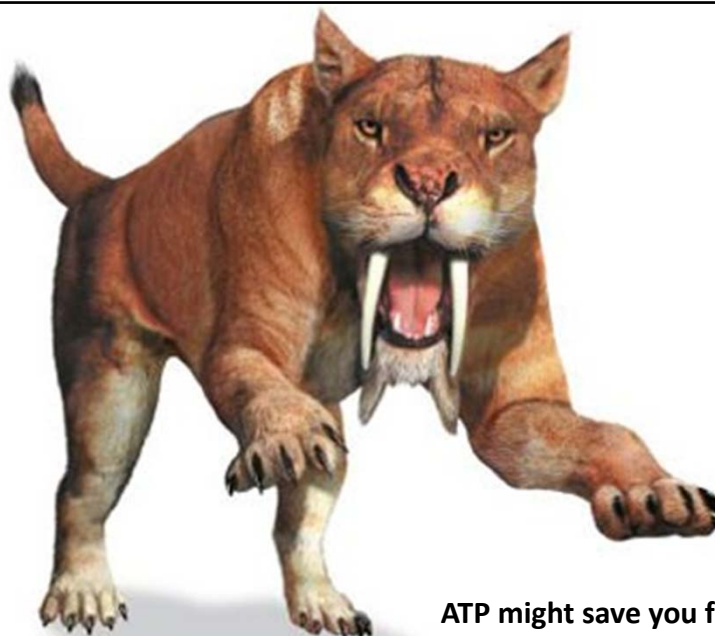
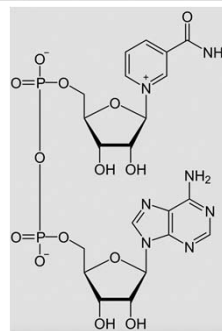


Critical metabolites

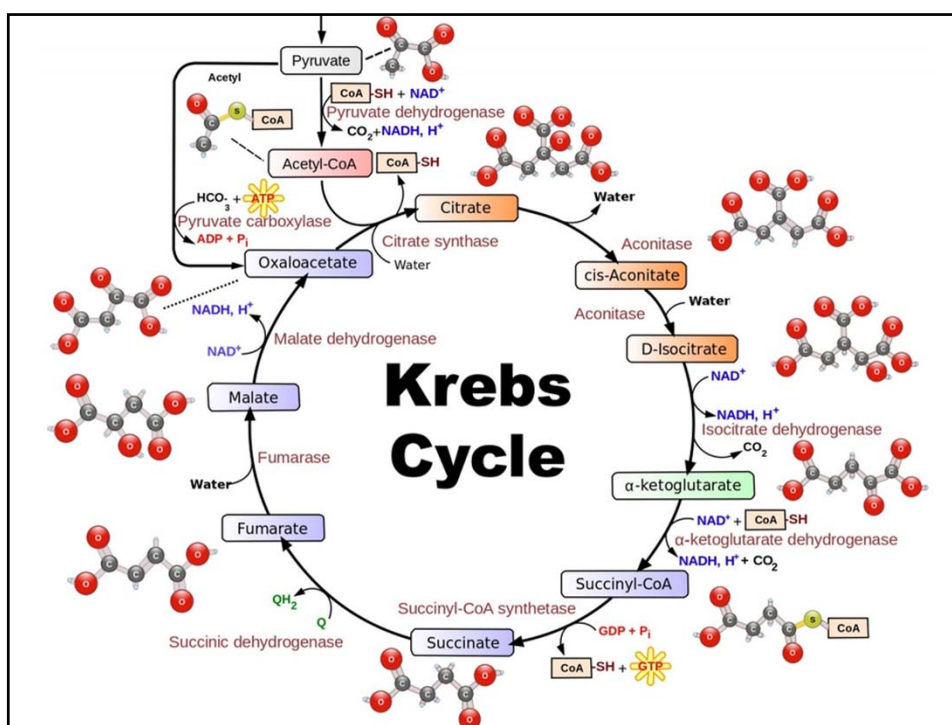
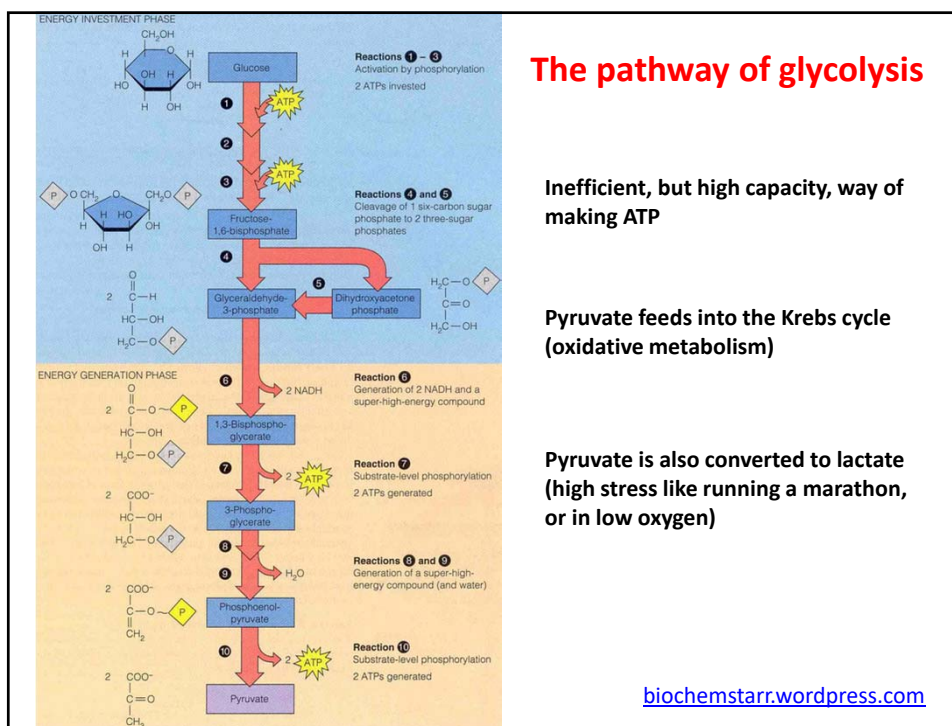
ATP: adenosine-5-triphosphate



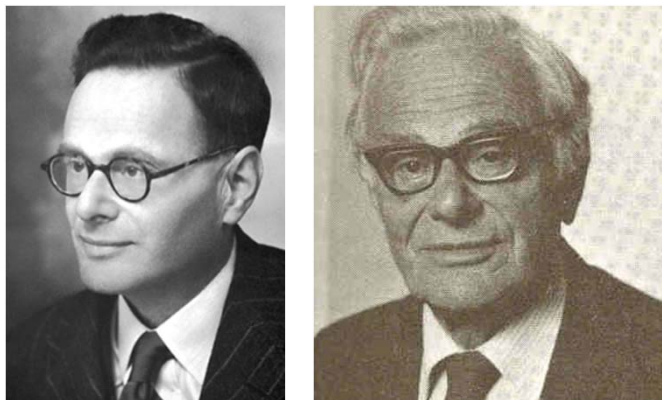
NAD⁺/NADH: nicotinamide adenine dinucleotide



**ATP might save you from
being his/her lunch!**

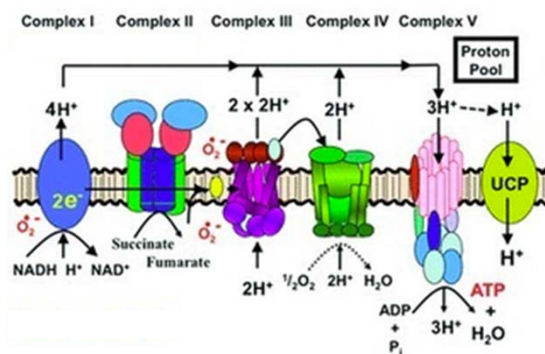


Sir Hans Krebs



Had the pleasure as a graduate student of introducing him at a seminar

Mitochondrial oxidative phosphorylation

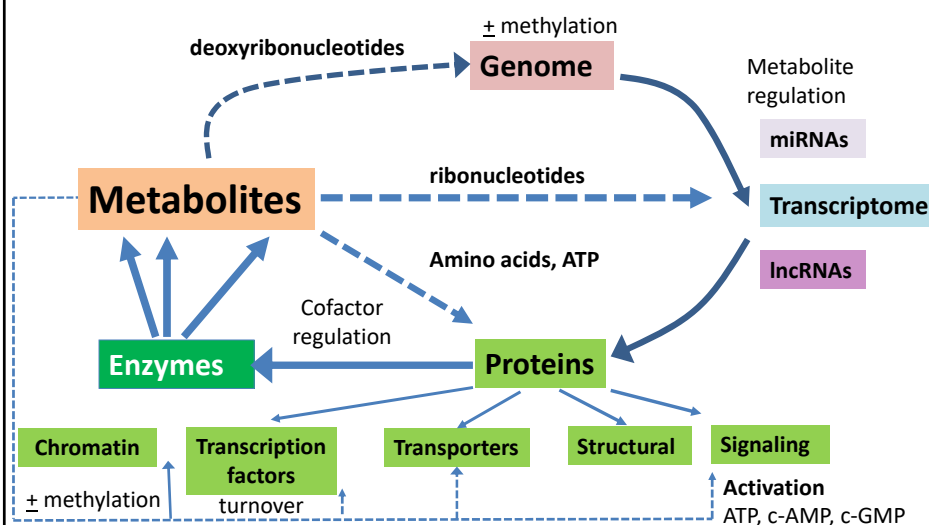


NADH from the Krebs cycle, as well as succinate, generate a proton (H^+) gradient (upper region) that drives rotation of one of the subunits of ATP synthase. This exposes the catalytic domain of this enzyme and makes ATP.

Understanding metabolites

- Metabolites represent the *action items* that come from gene expression and protein activity
 - They are found in the same range of concentrations as drugs
 - Metabolites (μM or mM) (acetyl CoA, ATP, SAdMet, αKG) are regulators of epigenetics
 - Bile acids (μM) are the natural ligands of FXR and LXR
 - Other metabolites (pM or nM) may be exquisite physiological regulators of kidney function (prostaglandins, F_2 -isoprostanes)
- Studying the metabolome requires multiple levels of science from the analytical to the physiologic to the computational

Metabolites are associated with every aspect of cellular events

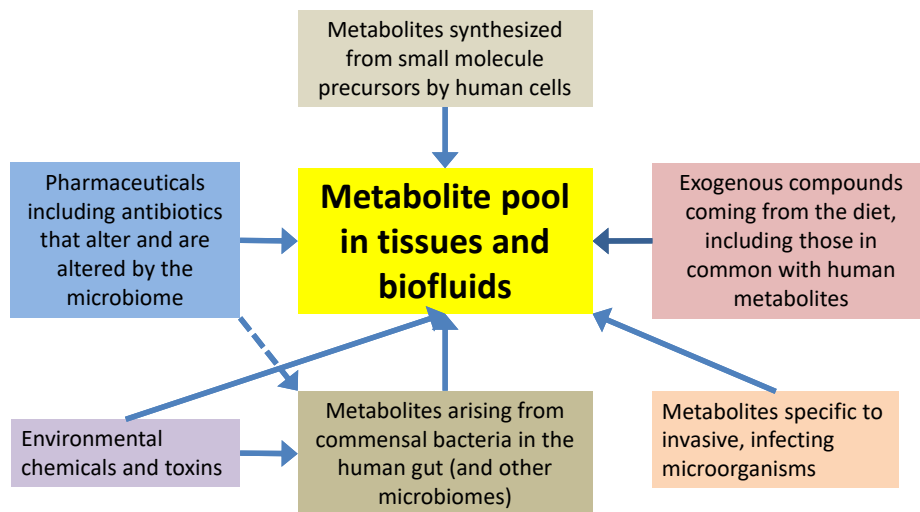


Metabolism and time

- Not only should metabolites appear in the right place, there is also the question of the importance of the timescale
- Metabolism defects in the heart may be only seconds away from death – rogue waves in metabolism??
- Irreversible damage to the brain may occur in minutes
- Go/No-Go decisions for a cell to divide or apoptose may occur in tens of mins



The metabolome is more than what's in textbooks



Overview of metabolome chemistry

Metabolites encompass an enormous range of chemistries

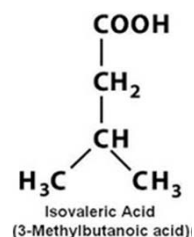
- **Gaseous**
 - H₂, H₂S
- **Volatile**
 - Butyric acid, acetone, skatole
- **Hydrophilic (water-loving)**
 - Glucose
- **Charged-positive/negative**
 - Amino acids, nucleotides, organic acids, amines
- **Hydrophobic (fat-loving)**
 - Lipids, steroids, hydrocarbons

Gases and volatiles

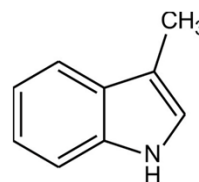
- **In breath**
 - H₂ from reductive anerobic bacteria
 - Lactose-intolerant
 - Measure of gut transit (typically 4-6 hours)
 - CO₂
 - From all carbon-containing substrates
 - From specific ¹³C-labelled substrates
 - Acetone (in diabetics)
 - Trimethylamine
 - From fish, or flavin monooxygenase (FMO3)-deficient subjects

Gases and volatiles

- **Sweat gland**
 - **Sweaty socks syndrome**
 - **Isovaleric acid** (leucine metabolism)
 - **Caused by bacteria or enzyme defect**



- **Flatulence**
 - **Mostly gases (H₂, CO₂ and H₂S), but with volatiles produced by colonic bacteria (**skatole**, from the amino acid tryptophan)**

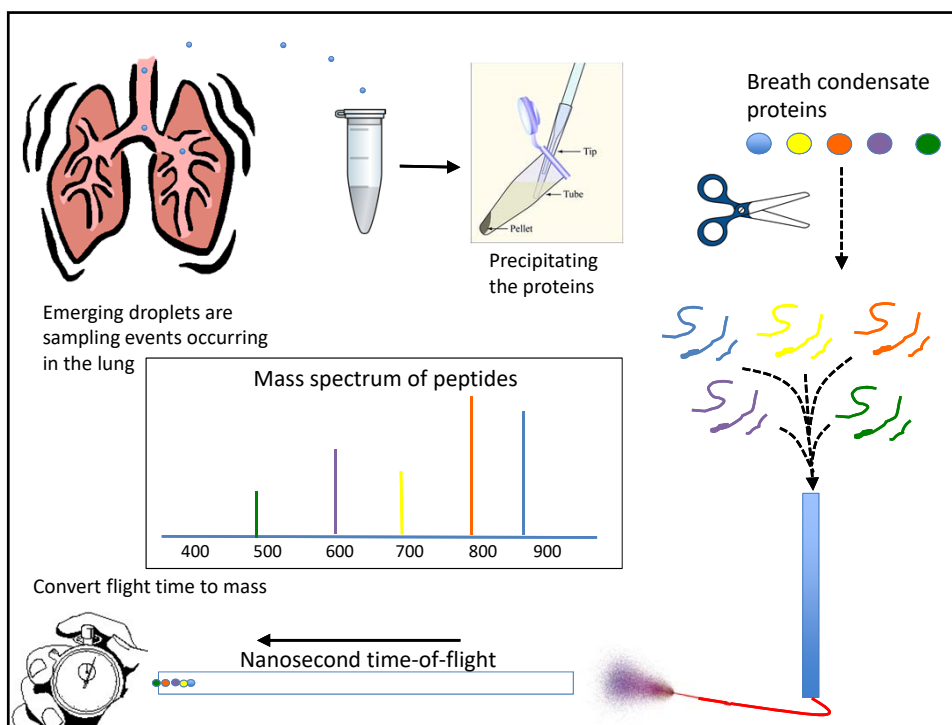


Other volatiles

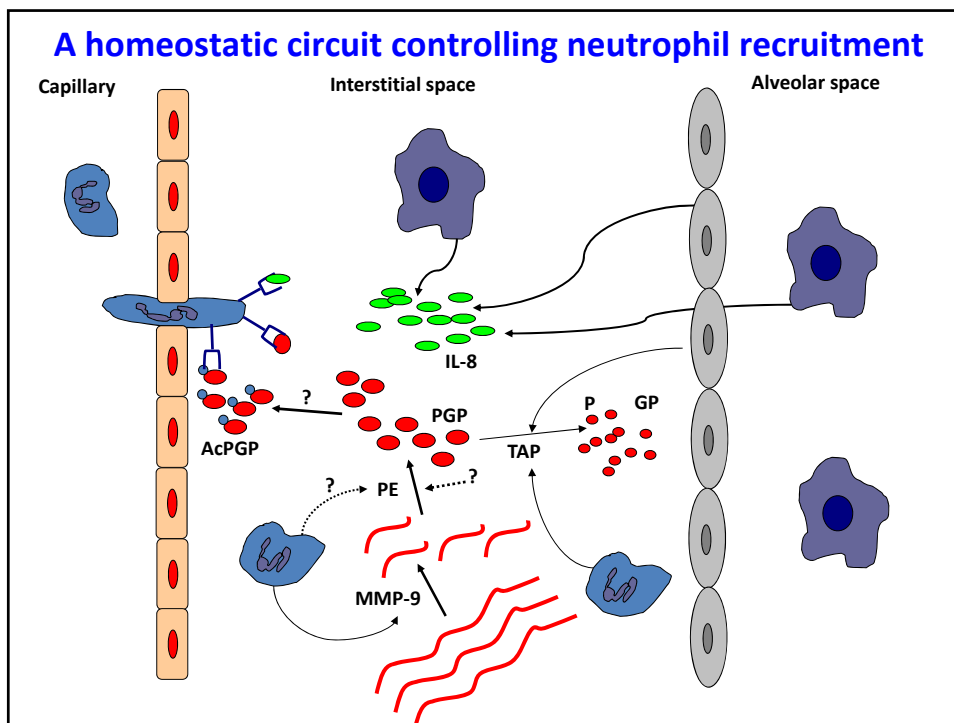
- **Short chain, unsubstituted fatty acids**
 - **Formic, acetic, propionic, butyric acids**
- **Will evaporate in the acidic form**
 - **Formic acid, b.p. 101°C**
 - **Acetic acid, b.p. 118°C**
 - **Propionic acid, b.p. 141°C**
 - **Butyric acid, b.p. 163.8°C**
 - **Isobutyric acid, b.p. 155°C**
- **Convert to ammonium salts before evaporating**

Breath condensates

- Not strictly consisting of volatiles
- A mist or spray created by the frothing of the fluids inside the lung
 - Condensable using a dry-ice cooled trap
 - Several ml of condensate can be easily collected in 5-8 min



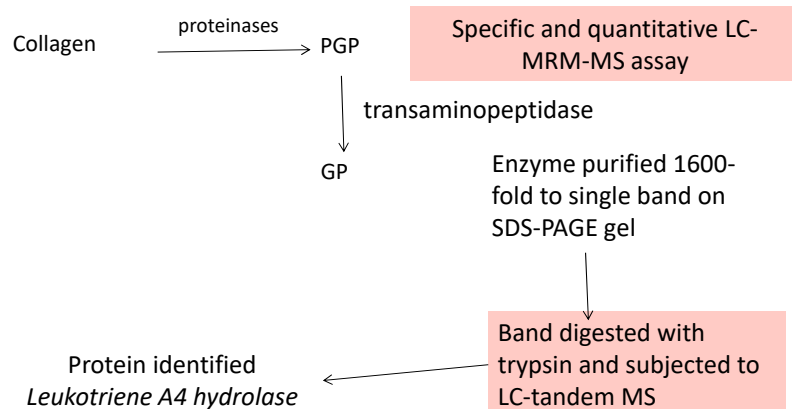
Certain metabolites are peptides



PGP is a common peptide in human collagen

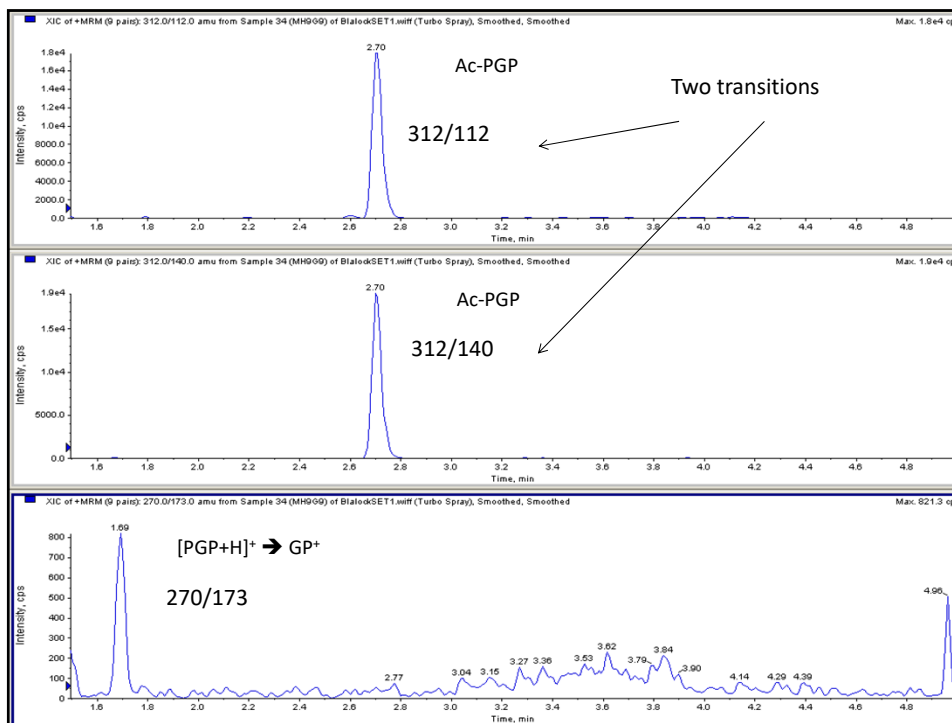
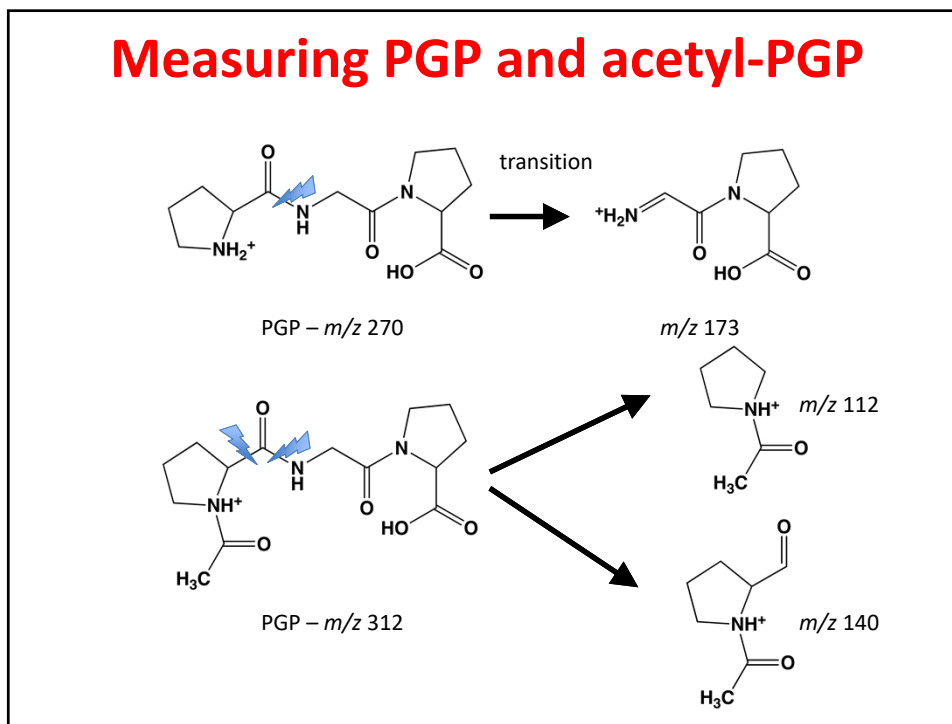
MFSFVDLRLLLLLAATALLTHGQEEGQVEGQDEDI PPIITCVQNGRLRYHDRDVWKPEPCRI
 CVCDNGKVLCDDDVICDETKNCPGAEVPEGECCPVC PDGSESP TDQETTGVVEGPKGDTGPR
 GPRGPAGPPGRDGI PGQPGLPGPP**PGPPGPPGPP**PGLGGNFAPQLSYGYDEKSTGGISV**PGP**
 MGPSGPRGL**PGPPGAPGP**QGFQGPPEPEGEPGASGPMGPRGP**PGPP**PGKNGDDGEAGKPGR
 PGERGP**PGP**QGARGLPGTAGLPGMKGHRGFSGLDGAKGDAGPAGPKGEPGSPGENGAPGQ
 MGPRGLPGERGRPGAP**PGP**AGARGNDGATGAAGP**PGP**TGPAGPPGFP GAVGAKGEAGPQGP
 RGSEGPQGV RGE**PGPPGP**AGAAGPAGNPGADGQPGAKGANGAPGIAGAPGFPGARGPSG
 QGPGGP**PGP**KGNSGEPGAPGSKGDTGAKGE**PGP**VGVQGP**PGP**AGEEGKRARGE**PGP**TGL
PGPPGERGGPSRGPFGADGVAGPKGPAGERGS**PGP**AGPKGSPGEAGRPEAGLPGAKGL
 TGSPGS**PGP**DGKTGP**PGP**AGQDGR**PGPPGP**PGARGQAGVMGF**PGP**KGAAGEPGKAGERGV
PGPPGAVGAPAGKDGEAGA QGP**PGP**AGPAGERGEQGPAGSPGFQGL**PGP**AGPPGEAGKPGE
 QGVPGDLGA**PGP**SGARGERGFPERGVQGP**PGP**AGPRGANGAPNDGAKDAGAPGAPGS
 QGAPGLQGM PGERGAAGL**PGP**KGDRGDAGPKGADGSPKDGVRGLTGPIGPPGAPAGPD
 KGESGSPGAPPTGARGAPGDRGE**PGPPGP**AGFAGPPGADGQPGAKGEPGDAGAKGDAGP
PGPAGPAGP**PGP**IGNVGAPGAKGARGSAGPPGATGFPGAAGRVP**PGP**SGNAGP**PGPPGP**
 AGKEGKGRGETGPAGRPGEVGP**PGPPGP**AGEKGS PGADGPAGAPGT**PGP**QGIAGQRGV
 VGLPQQRGERGFPL**PGP**SGEPGKQGPSGASGERGP**PGP**MGPPGLAGPPGESGREGAPGA
 EGSPGRDGS PGAKDRGETGPAGPPGAPGAPGA**PGP**VGPAGKSGDRGETGPAGPAGVGP
 VGARGPAGPQGP RDKGETGEQDRIKHRGFSGLQGP**PGP**PGSPGEQGPSGASGPAGP
 RGP PPSAGAPGKDGLNGL**PGP**IGP**PGP**RGRTGDAGPVGP**PGPPGPPGPPGPP**PSAGFDFS
 LPQPPQEK AHDGGRYYRADDANVVRDRDLEVDTT LKLSLQIENIRSPGSRKNPARTCR
 DLKMHSDWKSGEY WIDPNQGCNLDAIKVFCNMETGETCVYPTQPSVAQKNWYISKPNKD
 KRHVWFGESMTDGFQFEYGGQSDPADVAIQLTFLRLMSTEASQNIYHCKNSVAYMDQQ
 TGNLKKALLLQGSNEIEIRAEGNSRFTYSVTVDGCTSHTGAWGKTVEYKTTKTSRLPII
 DVAPLDV GADQEFDFDVG P VCFL

Mass spec contribution to PGP story



(Robert Snelgrove et al. Science, 2010)

Measuring PGP and acetyl-PGP



Metabolopeptidomics or peptidometabolomics

- **Are peptides metabolites?**
- **Are the tripeptides real?
Or is their mass simply coincident with the
empirical formula of another metabolite?**

Considering the case for tripeptides

- **Examine the basic physiology and
pharmacology**
- **Are there examples of bioactive tri-peptides?**
- **What about other oligopeptides?**
- **Where would they come from?**
- **Why does METLIN seem to always have tri-
and not other oligopeptides?**

**Tripeptides could come from foods,
but are hydrolyzed by peptidases in
the enterocyte to amino acids**

**Deficiencies in the peptidases could lead to
food and bacterial peptides entering the
systemic circulation**

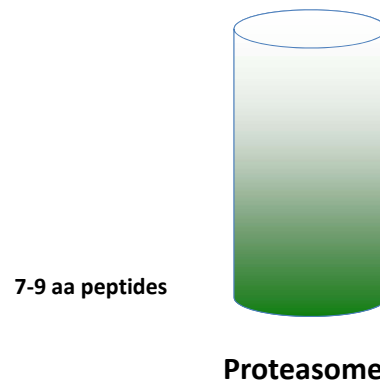
Can tripeptides have biological activity?

- **For toxicologists, there is one very familiar tripeptide without whom, I would not be giving this talk, or you to listen to it.**
- **Glutathione (GSH) – glutamyl-cysteinyl-glycine**
 - GSH reacts with free radicals to generate GSH conjugates and therefore protects many organs
- **It is synthesized from small molecule precursors**
 - However, it is a true metabolite, i.e., it is made from smaller precursors without the direct aid of ribosomes

Are there other sources of short peptides?

- Proteins undergo degradation in the proteasome caused by targeted ubiquitination
 - The digested products are peptides (escapees?)
- Lysosomes
- Autophagosome
- Neutrophil attack
- Other proteases (in renal tubules?)
- Foreign antigens hydrolyzed and presented on surface of cells

Protein \longrightarrow Protein-Ubq_n



Hydrophilic metabolites

- The most extreme hydrophilic metabolites without charged groups are the polyols:
 - Monosaccharides
 - Glucose
 - Fructose
 - Disaccharides
 - Lactose
 - Maltose
 - Oligosaccharides

Organic acids

- Besides the short chain fatty acids mentioned earlier, there are many organic acids representing important cellular pathways
 - Glycolytic intermediates
 - Glucose-1-P, Glucose-6-P, Fructose-6-P, Fructose-1,6-DP, Glyceraldehyde-3-P, Dihydroxyacetone-P, Glycerate-3-P, Phosphoenol-P, Pyruvate, Lactate
 - Krebs cycle
 - Citrate, cis-Aconitate, Iso-Citrate, α -ketoglutarate, Succinate, Fumarate, Malate, Oxaloactate and those resulting from pathway defects
 - Nucleotides
 - ATP, ADP, AMP, GTP, etc.

How could we isolate organic acids?

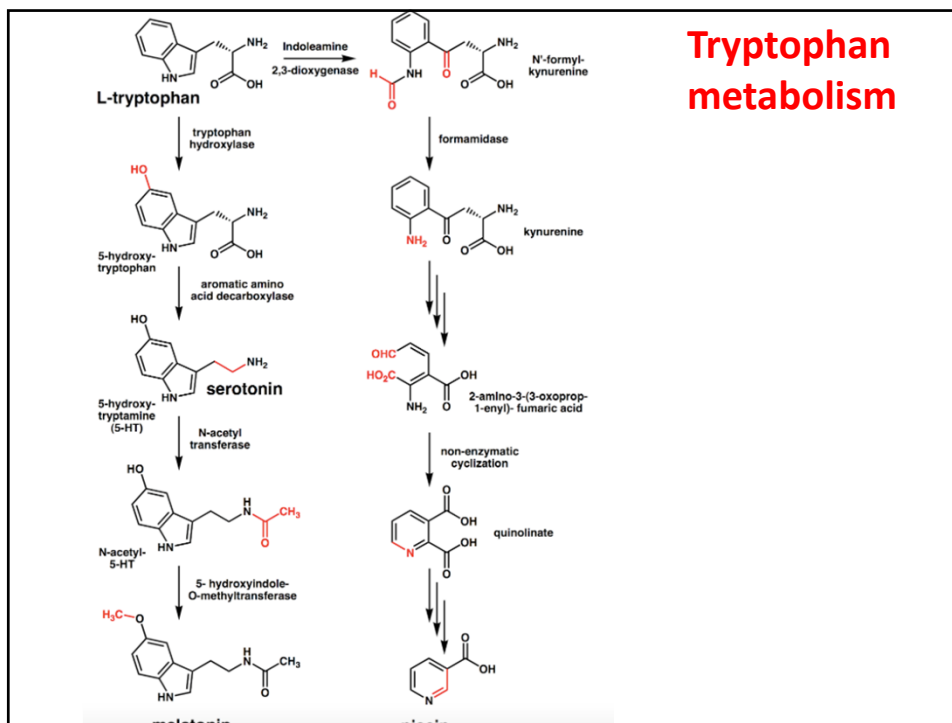
- Organic acids at neutral pH are negatively charged
- They will bind to anion exchange resins in say the formate form
- Can be eluted with ammonium formate or ammonium acetate (mass spec compatible)



AG-1

Amino acids

- All the α -amino acids found in proteins and their precursors and metabolites
 - Mostly L-isomers, but there are D-isomers in nature
- Tryptophan is oxidized to kynurenine and is a precursor to NAD(H) and NADP(H), serotonin melatonin and niacin
- β -Alanine is formed from uracil
 - 5-fluorouracil (anticancer drug) is converted to 2-fluoro- β -alanine which is in turn converted to bile acid conjugates



How could we isolate amino acids?

- Amino acids at neutral pH are positively charged
- They will bind to cation exchange resins in the H⁺ form



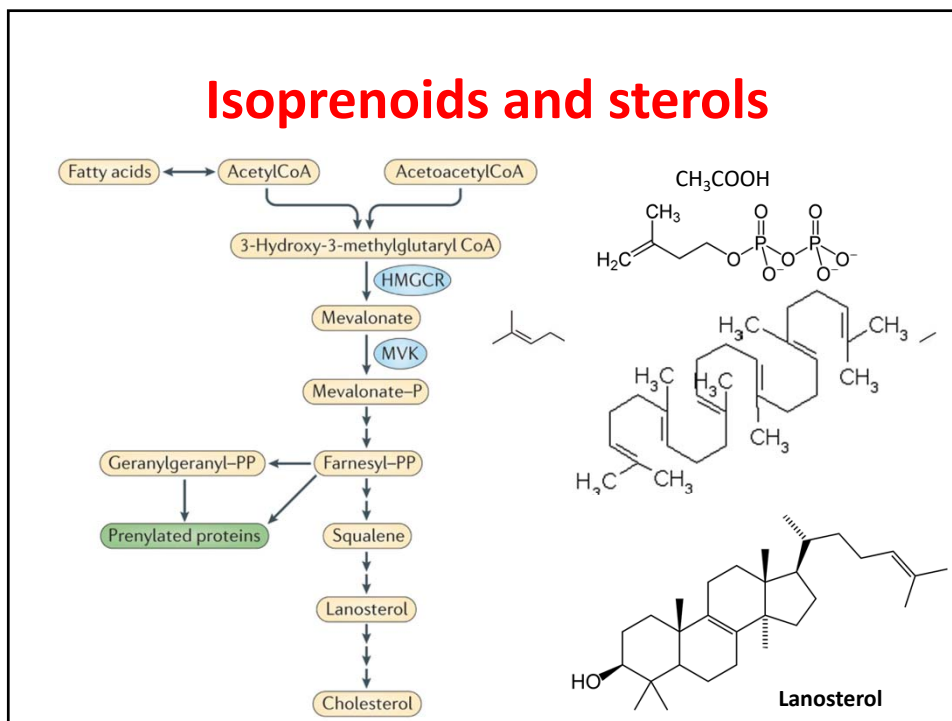
AG-50

- Can be eluted with ammonium hydroxide (mass spec compatible)

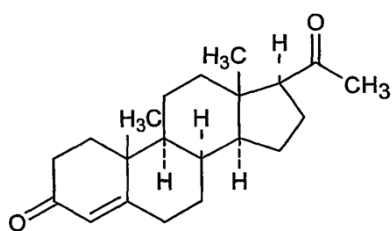
Hydrophobic metabolites

- These include sterols, steroid hormones, terpenoids, bile acids, vitamins A, D, E and K, and a vast array of lipids

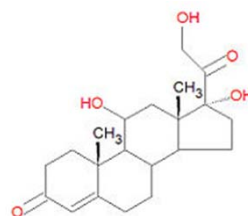
Isoprenoids and sterols



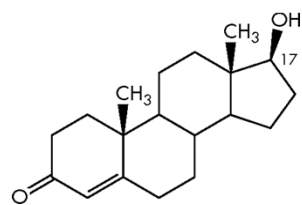
Steroids



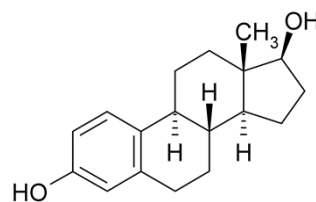
Progesterone



Cortisol

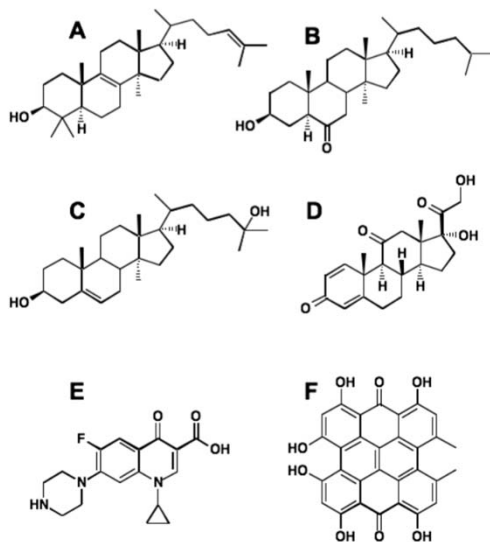


Testosterone



17β-estradiol

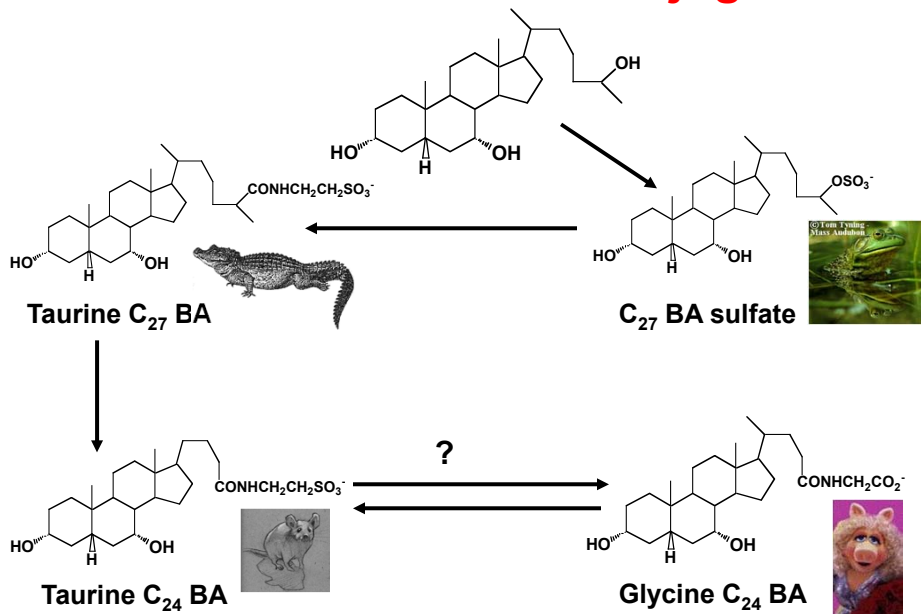
Importance of sterols and other compounds in lens cataracts



Structures A, B and C (all sterols) have recently been shown to have the property of “dissolving” lens cataracts. Cholesterol, on the other hand, has no effect. Other sterols observed in *cerebrotendinous xanthomatosis* promote cataracts.

D, E and F all promote lens cataracts. D is prednisone (an anti-inflammatory steroid), E is ciprofloxacin (an antibiotic) and F is hypericin from the botanical, St. John’s wort.

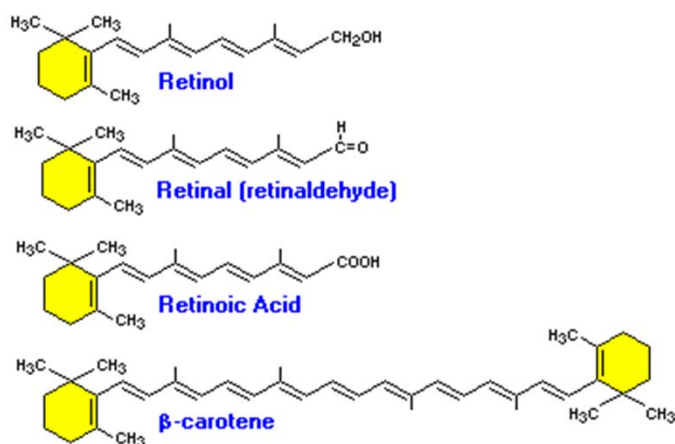
Evolution of bile acid conjugation



The vitamins

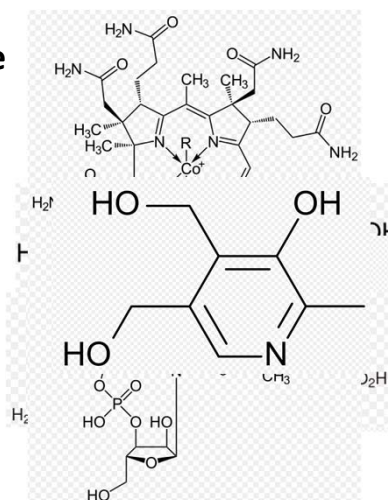
Lack of these leads to serious illness, but not death

Vitamin A



Vitamin B

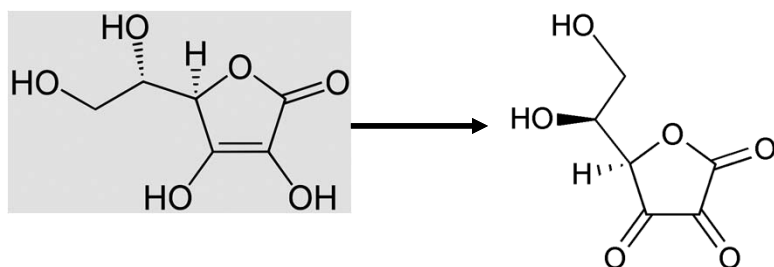
- They are all water-soluble
 - Vit B₁ – thiamine
 - Vit B₂ – riboflavin
 - Vit B₃ – niacin
 - Vit B₅ – pantothenic acid
 - Vit B₆ – pyridoxine
 - Vit B₇ – biotin
 - Vit B₉ – folic acid
 - Vit B₁₂ – cobalamins



They are not made by human enzymes and if deficient in the diet cause disease

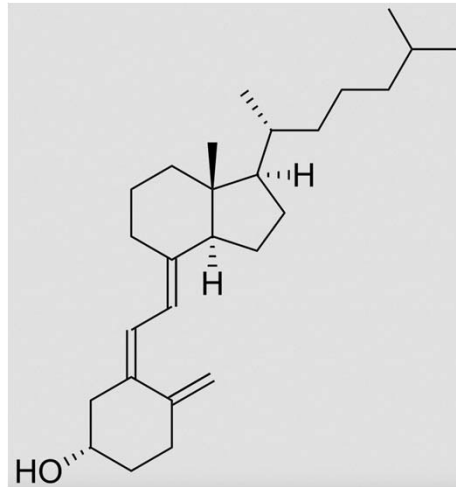
Vitamin C

- Ascorbic acid



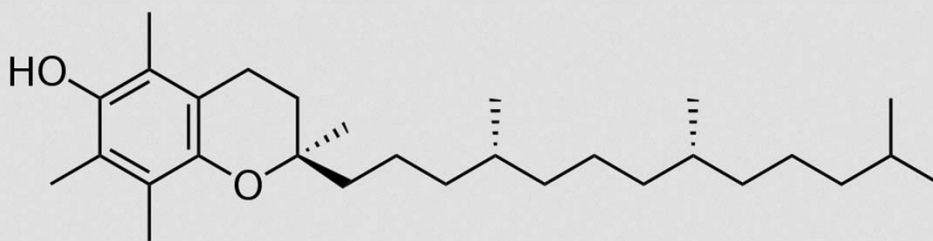
dehydroascorbic acid

Vitamin D



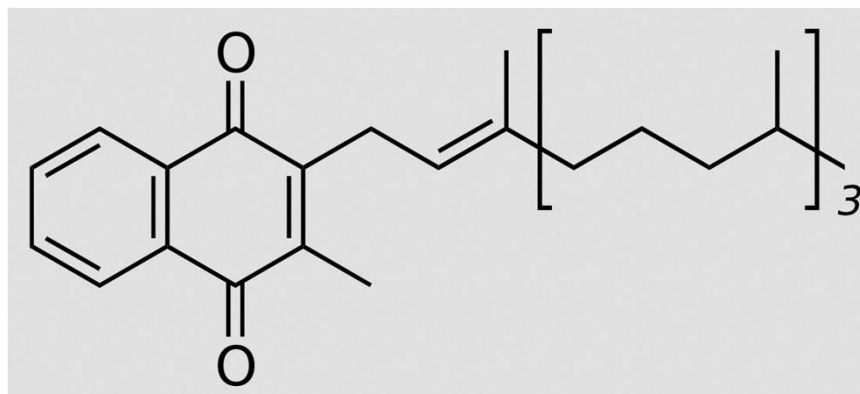
In fish, supplemented in milk, made in skin by UV light

Vitamin E



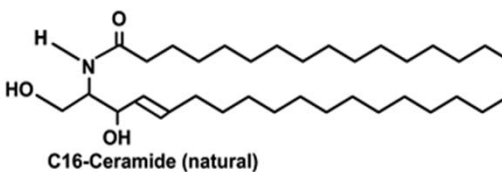
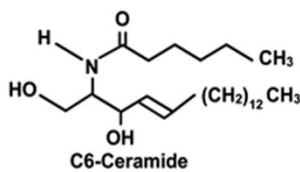
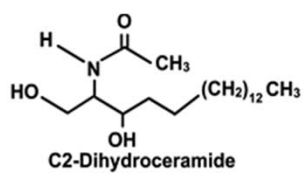
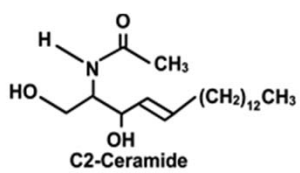
Found in oils from plants

Vitamin K

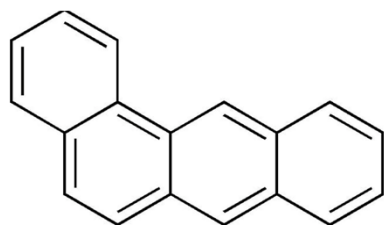


Is an anticoagulant – needed to stop bleeding

Phospholipids



Hydrocarbons



Benz[a]anthracene
In smoke from barbecued meat



Cetyl palmitate
In hair shampoo

Solubilities of the metabolites

- Those in biological fluids are “in solution”, but may not be soluble in water or methanol alone
 - Are glucose or amino acids soluble in methanol?
 - Are cholesterol esters in plasma soluble in methanol or water?
 - If a metabolite binding protein is precipitated by methanol, does the metabolite still bind to it?
 - Does pH have an effect on solubility?